

QUESTION 1

- (a) A statistical mechanics treatment of molecules of a monatomic gas is usually made in terms of **microstates** and **macrostate** of the system. Explain the meanings of the highlighted terms in this statement. State the connection between them.
- (b) Consider an ideal monatomic gas sample with molecules of mass, m , placed in a cubic box of side, L . A molecule of the gas would have velocity components, v_x , v_y , and v_z , respectively, in the Cartesian coordinate directions, x , y , and z . For a typical coordinate, x , say,
- Write down an expression for the momentum, p_x .
 - Use Heisenberg uncertainty principle to deduce an expression for the corresponding energy E_x for this coordinate, in terms of an assigned quantum number, n_x , and the dimension of the box.
 - Hence write down expressions for the total energy of the molecule.
 - Given that the box is of side 10 cm, estimate a typical value for the number of quantum states, n , of the system, at temperature 300 K. {Hint: {You may use the postulate of **equal a priori probability distribution**, and the value of Boltzmann constant}.

QUESTION 2

- (a) Define the term **Partition Function**, used in the study of statistical mechanics. Using suitable examples, discuss the importance of the Partition Function in the study of statistical mechanics.
- (b) The translational Partition Function, Z , for one molecule of an ideal monatomic gas in a container, is given by

$$Z = V \left[\frac{2\pi m k_B T}{h^2} \right]^{3/2}$$

- (c) State what the symbols stand for in the above expression.
- (d) Derive an expression for $\log_e Z$

(e) Estimate the numerical value for the above expression for 2 moles of helium gas in a container

QUESTION 3

- (a) In statistical mechanics we can treat a given system consisting of collections of **distinguishable**, or **indistinguishable** particles. Explain the meanings of the highlighted terms. Give one example of systems falling in each category.
- (b) Explain the term **degeneracy**, used in the study of a system of indistinguishable particles in statistical mechanics.
- (c) The total energy, E , of a hypothetical system of indistinguishable particles is given by $E^2 = n^2 h^2 f^2 = 66 h^2 f^2$, where the symbols have their usual meanings, n being the number of quantum states, which can have components n_x , n_y , and n_z , along the respective coordinate axes. Copy Table 1 below in your answer booklet. Using the lead provided in the first column, complete the table by finding values of n_x , n_y , and n_z that satisfy the above energy relationship. Hence deduce the degeneracy of this hypothetical energy state.

Table. 1

$$E^2 = n^2 h^2 f^2 = 66 h^2 f^2,$$

n_x	7														
n_y	1														
n_z	4														

deduce a numerical value of the entropy S for 2 moles in a container or

SECTION B: Answer any TWO questions

1(a) Explain what you understand by the Ultraviolet Catastrophe.

(b) Four particles are to be distributed among four energy levels $\epsilon_1 = 1, \epsilon_2 = 2, \epsilon_3 = 3, \epsilon_4 = 4$ units having degeneracies $g_1 = 1, g_2 = 2, g_3 = 2, g_4 = 1$ respectively. The total energy of the system is 10 units. Find the possible distribution (macrostates) and the microstates corresponding to the most probable macrostates. Assume that the particles are: (i) distinguishable (ii) indistinguishable bosons (iii) indistinguishable fermions.

2(a) Explain the contributions of Wien, Raleigh and Jeans in the development of Blackbody radiation theory.

(b) A system has non-degenerate single-particle states with 0,1,2,3 energy units. Three particles are to be distributed in these states such that the total energy of the system is 3 units. Find the number of microstates if the particles obey (i) MB statistics (ii) BE statistics (iii) FD statistics. Find the corresponding macrostates and microstates also.

3(a) Briefly explain the following:

(i) Maxwell Boltzmann Distribution

(ii) Bose- Einstein Statistics

(iii) Fermi-Dirac Statistics

(b) Three distinguishable particles ($N = 3$) (Yellow, Violet, Green) are to be distributed in a 3-level system ($l = 3$) where first and second levels are non-degenerate but the second level is doubly degenerate $\{g_1, g_2, g_3\} = \{1, 2, 1\}$. Identify all possible microstates for the system when the occupancy sequence of the system is $\{n_1, n_2, n_3\} = \{1, 1, 1\}$. *